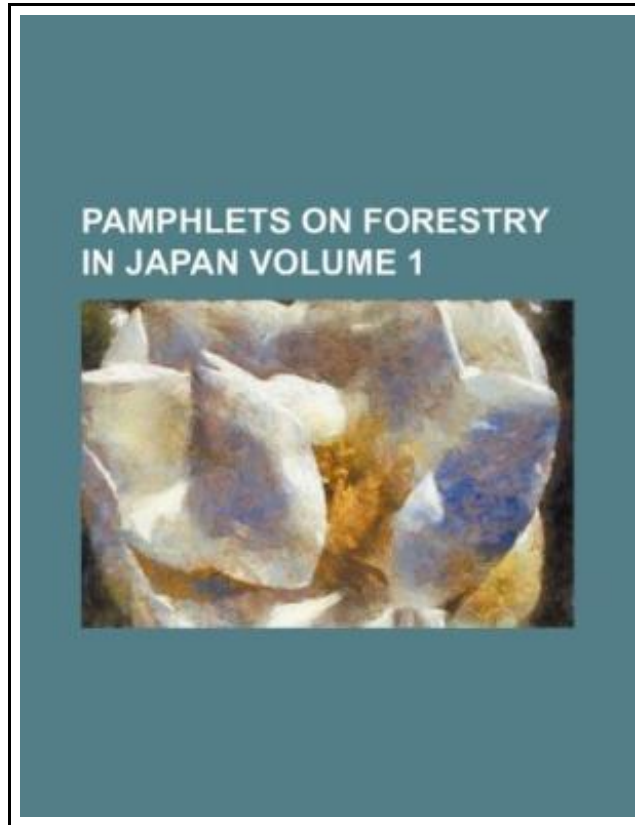


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1904 Excerpt: .above obtained was established by the hydrochloride of the latter (CH₃OH,6H Cl). It is noteworthy that the melting point of pinene hydrochloride was 132.5-133.5C. in my experiment (Corr), This is different from the observation made by Wallach who puts the melting point of the same substance at 125C. (Ann. Chemie, 239, 4) and J. H. Long s report of 131C. as the melting point (Jour of Amer. Chem. Soc. 21, 637). The third fraction (240-270) gave a slight yellowish liquid of a somewhat visicous nature possessing the wood smell of Hinoki. After the liquid was rectified by distillation over sodium under atmospheric pressure the chief fraction was collected between 260-270C. The oil thus separated boils mostly between 267-268C. When the fraction is dissolved in a little glacial acetic acid and is shaken with a few drops of concentrated sulphuric acid, the liquid assumes a faint reddish color which turns to red with a violet tint after a little while. After standing over night, it finally turns to brownish red. The identification of cadinen with this fraction was confirmed by its hydrochloride (C₁₆H₂₄. 2 HCl, m. p. 117-118C.) and hydrobromide (C₁₆H₂₄. 2 HBr, m.p. 124-125). Again, the rectified oil of the original product was distilled under atmospheric pressure in a flask provided with Le-Bel Henninger s stillhead; the fraction distilling under 170C. was collected to the amount of 60 a of the original oil; the substance thus collected is colorless and free from empyreumatic odour. The physical properties...



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